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Applications of the Third Law of Thermodynamics. An important application of the third law of thermodynamics is that it helps in the calculation of the absolute entropy of a substance at any temperature 'T'. These determinations are based on the heat capacity measurements of the substance.

### Third Law of Thermodynamics - Entropy at Absolute Zero

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Entropy is a quantity in thermodynamics that measures the disorder in a system. An object or substance with high entropy is highly disordered. Entropy in the universe can only increase. The third law of thermodynamics says that the entropy of a perfect crystal at absolute zero is exactly equal to zero.

### The Third Law of Thermodynamics & Its Application to ...

Web page for Thermodynamics and Its Applications, by Jefferson W. Tester. Thermodynamics and Its Applications Jefferson W. Tester and Michael Modell: Download current updated errata for the textbook View the textbook table of contents Download answers to selected problems E-mail the authors.

### Thermodynamics and Its Applications

What are the applications of the second law of thermodynamics? 1) According to the law, heat always flows from a body at a higher temperature to a body at the lower temperature. This law is applicable to all types of heat engine cycles including Otto, Diesel, etc. for all types of working fluids used in the engines.

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**What is the practical use of the Third Law of Thermodynamics?**

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The second law of thermodynamics is considered to be the most fundamental law of science. It explains not only the working of engines, refrigerators and other equipments used in our daily life, but also highly advanced theories like big bang, expansion of universe, heat death etc. Let us see applications of second law of thermodynamics to automobiles and refrigerators.

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